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Mineral chemistry of late Variscan gabbros from central Spain: constraints on crystallisation processes and nature of the parental magmas

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Abstract

Major and trace element chemistry has been determined in the main minerals (olivine, orthopyroxene, clinopyroxene, plagioclase, amphibole, phlogopite) from the Talavera and La Solanilla gabbroic intrusions (Spanish Central System). The results show a heterogeneous mineral composition illustrative of the chemical evolution of calc-alkaline basic magmas during crystallization. The formation of incompatible element-rich interstitial minerals, such as amphibole, phlogopite and accessory phases (e.g., zircon, apatite), and the presence of zoning patterns in major minerals towards a more evolved composition, point to a progressive differentiation of the parental melt. The chemical variation depicted by trace elements usually incompatible with clinopyroxene (e.g., rare earth elements (REE), Th, Zr, Nb, Ti) implies co-precipitation with phases displaying a compatible behaviour with respect to these elements (e.g., apatite, zircon, amphibole). However, the very high LILE (large ion lithophile elements) and LREE (light REE) shown by clinopyroxene can not be explained exclusively by a closed-system in-situ crystallization process. Several features of the mineral chemistry (e.g. positive correlation of (La/Sm)_N with La and Na in clinopyroxene and plagioclase, respectively) support the involvement of an external component (wall-rock assimilation or hybridization with an evolved melt). The petrography and mineral chemistry of the gabbros, in conjunction with bulk rock data from previous studies, favour a calc-alkaline affinity for the parental melts. Thus, metasomatism in the mantle sources related with this basic magmatism might be associated with recycling of crustal components during the Variscan orogeny.

Keywords: Late orogenic gabbros, late-stage crystallization, Spanish Central System, Variscan Orogeny, calc-alkaline magmas

Resumen

Se ha caracterizado la composición química de elementos mayoritarios y trazas en los minerales principales de las intrusiones gabroideas de Talavera y La Solanilla (Sistema Central Español). Los resultados muestran una composición mineral heterogénea que es ilustrativa de una evolución química de los magmas básicos calco-alcalinos durante la cristalización. La formación de minerales intersticiales con elevados contenidos en elementos incompatibles, como por ejemplo el anfíbol, la flogopita y algunos minerales accesorios (circón, apatito...), y la presencia de zonados en los minerales principales, con tendencia a desarrollar una composición más evolucionada hacia el borde, indican la diferenciación progresiva del fundido parental. La variación química de elementos traza normalmente incompatibles en el clinopiroxeno (e.g., REE, Th, Zr, Nb, Ti) implica la co-precipitación de fases con un comportamiento compatible con dichos elementos (e.g., apatito, circón, anfíbol). Sin embargo, las elevadas concentraciones de LILE y LREE del clinopiroxeno no se pueden explicar exclusivamente por un proceso de cristalización en un sistema cerrado. Varias características de la química mineral (e.g., correlación positiva de la relación (La/ Sm)_N con el contenido de La y Na en el clinopiroxeno y la plagioclasa, respectivamente) suponen la participación de un componente externo (asimilación de roca encajante o mezcla con un fundido más evolucionado). La petrografía y la química mineral de los gabros, junto con los datos de roca total de estudios previos, favorece una afinidad calco-alcalina para los fundidos parentales. Por lo tanto, el metasomatismo en las fuentes de manto asociadas con este magmatismo básico podría estar relacionado con el reciclaje de componentes corticales durante la orogenia Varisca.

Palabras clave: Gabros tardiorogénicos, Sistema Central Español, Orogenia Varisca, magmas calco-alcalinos

1. Introduction

During cooling of magmatic bodies, re-equilibration of interstitial melts with early-formed mineral assemblages is a key process which might influence the final bulk rock composition (e.g. Barnes, 1986). The relevance of late-magmatic processes in the geochemical evolution of magmas has been highlighted by many studies focused on mafic intrusions from oceanic (e.g. Ross and Elthon, 1997; Tribuzio et al., 1999b, 2000; Natland and Dick, 2001; Gao et al., 2007) and continental (e.g. Cawthorn et al., 1992; Bédard, 1994; Tribuzio et al., 1999a; Claeson and Meurer, 2004; Tiepolo and Tribuzio, 2005) environments. These studies have shown that in-situ crystallization of evolved interstitial melts directly controls the final mineral composition, and may result in enrichment of incompatible trace elements in the whole-rock composition. In accordance with these observations, a variable proportion of interstitial minerals, including clinopyroxene, H₂O-bearing phases (amphibole, phlogopite) and other accessory minerals (e.g. apatite, zircon, ilmenite, baddelevite) are commonly formed between early-precipitated crystals (e.g. Tribuzio et al., 1999a, b, 2000; Tiepolo and Tribuzio, 2005; Borghini and Rampone, 2007). Late overgrowth of the liquidus phases may also occur (e.g. Barnes, 1986; Borghini and Rampone, 2007). Even though most of the above studies deal with cumulate rocks, the processes they discuss for the final stages of crystallization, may also operate during solidification of non-cumulate mafic rocks.

Another important aspect is the possible interaction between early crystallized minerals and a variable proportion of evolved interstitial liquid. The outcome of this process will be very different in a "closed system" crystallization process (e.g. Barnes, 1986; Bédard, 1994; Charlier et al., 2005; Bernstein, 2006) and if an external component, such as an evolved melt associated with wall rock assimilation or felsic magma hybridization, plays a major role (e.g. Tribuzio et al., 1999a; Tiepolo and Tribuzio, 2005; Renna et al., 2006). Mineral-melt interaction during late stage crystallization has been identified in rocks of variable geochemical affinity, including tholeiitic (Gillis, 1996; Tribuzio et al., 1999b, 2000; Coogan et al., 2001) and continental calc-alkaline magmas, the latter sometimes associated with post-collisional geodynamic settings (Monjoie et al., 2005; Tiepolo and Tribuzio, 2005; Renna et al., 2006). The nature of magmas formed during orogenic events is commonly unclear due to the involvement of several geochemical reservoirs (crust and mantle) and the leading role played by differentiation processes (e.g., fractional crystallisation and crustal contamination), which may obliterate the primary geochemical signature.

The determination of major and trace element mineral composition by microanalytical techniques (Electron Microprobe and Laser Ablation ICP–MS) in rock suites subjected to complex crystallization stories has proved as a main tool in deciphering the type of operating magmatic processes (e.g., Tribuzio *et al.*, 1999b; Coogan *et al.*, 2001; Claeson

and Meurer, 2004; Bernstein, 2006; Borghini and Rampone, 2007). This approach allows the characterization of trace element mineral zoning and other micro-scale chemical heterogeneities.

In the present study we have analysed the major and trace element composition of the main rock-forming minerals from two gabbroic intrusions cropping out in the Spanish Central System (SCS) (Variscan Iberian Belt). These Variscan gabbros have been previously studied with special attention focused on the whole-rock chemistry (e.g. Orejana et al., 2009; Scarrow et al., 2009), the major element mineral data (Molina et al., 2009), and their zircon isotope (U-Pb, Hf) geochemistry (Villaseca et al., 2011). Different hypothesis have been proposed for their origin, advocating either primary calc-alkaline magmas (Orejana et al., 2009) or alkaline mantle-derived melts mixed with felsic magmas at the crust-mantle boundary (Scarrow et al., 2009). Our study represents the first attempt to determine the trace element mineral composition of SCS gabbroic rocks, especially regarding the chemical heterogeneity and zoning of interstitial phases. This kind of mineral data makes it possible to characerize late magmatic processes and help constrain the nature of the parental magmas.

2. Geological Setting

The Spanish Central System (SCS) batholith is mainly formed by more than 100 granitic intrusions (e.g. Bea et al., 1999; Villaseca and Herreros, 2000) emplaced within Neoproterozoic to Palaeozoic metaigneous and metasedimentary series, which represent the innermost continental region of the Iberian Variscan Belt (Fig. 1). Although these intrusions are predominantly peraluminous monzogranites, several series have been distinguished, including S-type peraluminous granites, I-type metaluminous to peraluminous plutons and transitional peraluminous granitoids (e.g. Recio et al., 1992; Pinarelli and Rottura, 1995; Villaseca et al., 1998; Villaseca and Herreros, 2000). Several models have been proposed to explain the origin of this huge batholith: a) hybridization between crustal melts and mantle-derived magmas (Pinarelli and Rottura, 1995; Moreno-Ventas et al., 1995); b) variable degrees of crustal assimilation by mantle derived magmas (Ugidos and Recio, 1993; Castro et al., 1999); and c) partial melting of essentially crustal sources, either from lower crustal derivation (Villaseca et al., 1998, 1999) or from midcrustal levels (Bea et al., 1999, 2003). Clear geochemical similarities between SCS granites and lower crustal xenoliths carried by the SCS Upper Permian alkaline dykes, suggest that the lower crust is the most likely crustal source for the formation of the SCS batholith (Villaseca et al., 1999, 2007). This is reinforced by coincidence in the U-Pb zircon age ranges recorded in the SCS granites and the lower crustal xenoliths (Fernández-Suárez et al., 2006; Orejana et al., 2011).

Basic magmatism in the SCS batholith is represented by several series of fairly different geochemical affinity (calc-



Fig. 1.- Geological map of the Spanish Central System showing the location of the studied Variscan gabbros (Talavera and La Solanilla), modified from Franco and García de Figuerola (1986), Franco and Sánchez García (1987), and Orejana *et al.* (2009).

alkaline, alkaline and tholeiitic) (Fig. 1), which intruded from Carboniferous to Jurassic (e.g. Villaseca *et al.*, 2004). The alkaline and tholeiitic mafic rocks are post-Variscan dyke swarms associated with a rifting regime prior to the Atlantic Ocean opening (e.g. Cebriá *et al.*, 2003; Orejana *et al.*, 2008). The calc-alkaline gabbroic intrusions, which are mainly concentrated in the western SCS (Franco and García de Figuerola, 1986; Franco and Sánchez–García, 1987), are small (usually < 1 km in length) Variscan post-tectonic plutons with ages mainly in the range 305–300 Ma (Zeck *et al.*, 2007; Villaseca *et al.*, 2011). According to recent data, they are coeval with the main SCS granitoid intrusions (~308–298 Ma; Zeck *et al.*, 2007; Díaz–Alvarado *et al.*, 2011; Orejana *et al.*, 2012).

Recent studies have focused their attention on the nature and petrogenesis of the SCS mafic bodies (Molina *et al.*, 2009, 2012; Orejana *et al.*, 2009; Scarrow *et al.*, 2009; Villaseca *et al.*, 2011). These gabbros exhibit high incompatible trace element concentrations (LILE, LREE, Pb), in conjunction with negative Nb–Ta–Ti anomalies, and a heterogeneous Sr–Nd isotopic composition (e.g. Orejana *et al.*, 2009; Scarrow *et al.*, 2009). As a result, the most primitive gabbros have been interpreted as derived from melting of a heterogeneous subcontinental lithospheric mantle, metasomatised via crustal recycling (Orejana *et al.*, 2009). However, other studies favour a mixing process in the deep crust involving alkaline mantle-derived magmas and crustal melts (e.g. Scarrow *et* *al.*, 2009). In any case, subduction is not considered to be directly linked with the genesis of the gabbros, and melting would have been caused by adiabatic decompression after the collision ended (Orejana *et al.*, 2009; Scarrow *et al.*, 2009).

Additionally, hybridization of these basic magmas with crustal melts during transport and emplacement has been described by Moreno–Ventas *et al.* (1995), Scarrow *et al.* (2009) and Villaseca *et al.* (2011). Other studies on Iberian Variscan gabbros have also highlighted the existence of crustal contamination at shallow levels on the basis of whole-rock chemistry (e.g. Dias and Leterrier, 1994; Galán *et al.*, 1996; López–Moro and López–Plaza, 2004). More recently, it has been observed that differences in initial eHf values between zircon crystals from several gabbroic bodies are in accordance with the heterogeneity of whole-rock Sr–Nd isotopic data, and that within sample eHf variability also indicates that crustal assimilation and felsic magma hybridization occurred in some of these intrusions (Villaseca *et al.*, 2011).

The two gabbroic bodies studied for the present study (Talavera and La Solanilla) are hosted by granitic and metamorphic rocks. Contact relations with the wall-rock can not be clearly observed in the field. Samples collected from different outcrops in the inner areas are fairly homogeneous in terms of bulk-rock (Orejana *et al.*, 2009) and mineral composition. The main gabbro bodies do not show modal layering or textural variations. This homogeneity and their primitive character, make these

Sample	Gabbroic massif	UTM coordinates (Zone 30T)	Early crystallization phases*	Inerstitial phases*	Accessory phases*	Modal composition**	Representative lithology and textures			
107038	Talavera					ol (7%), pl (54%),	Medium- to coarse-grained olivine- bearing gabbronorite.			
107038L	Talavera	x: 343.136 y: 4.429.390	.1 .1	cpx, Na-pl, amph, phl, ilm	ilm, zrn, ap, ttn, bdy	opx (16%), cpx (12%),	Intergranular texture with anhedral			
107039	Talavera		01, p1, 0px			amph (6%), phi (3%) and other accessory	poikilitic orthopyroxene.			
107042	Talavera					phases (2%)	Extensive transformation of interstitic clinopyroxene by amphibole.			
108585	La Solanilla					a1(15%) $n1(58%)$	Medium-grained olivine-bearing			
108585L	La Solanilla	x: 297.511	11	cpx, opx, Na-	ilm, ap, ttn, zrn, bdy	opx (6%), cpx (14%),	gabbronorite. Intergranular texture with anhedral			
108588	La Solanilla	y: 4.478.590	p1, 01	ilm		amph (3%), phi (3%) and other accessory	olivine and prismatic plagioclase. Amphibole-orthopyroxene coronas			
108591	La Solanilla					phases (1%)	around olivine.			

*ol: olivine, opx: orthopyroxene, cpx: clinopyroxene, pl: plagioclase, Na-pl: Na-rich interstitial plagioclase, amph: amphibole, phl: phlogopite, ilm: ilmenite, zrn: zircon, ap: apatite, ttn: titanite, bdy: baddeleyite.

**Modal composition has been estimated by point counting performed on thick sections nº 107038L and 108585L (see the analytical methods for more details).

Table 1.- Summary of petrographic features of analysed SCS olivine-bearing gabbros

gabbros suitable for the study of the magmatic processes which operated during cooling of the parental basic magma.

3. Analytical methods

Eight samples were selected for the present study: four from the Talavera gabbro and four from the La Solanilla gabbro (Fig. 1; Table 1). Thin sections were used for petrographic observations and for a detailed determination of the mineral mode. The latter was done statistically considering the relative abundance of the main minerals within a virtual network of 1000-2000 points (depending on the sample surface) (see results in Table 1). Major element compositions of olivine, orthopyroxene, clinopyroxene, plagioclase, amphibole and phlogopite has been analysed directly on polished thin and thick (130 µm) sections at the Centro Nacional de Microscopía Electrónica (Complutense University of Madrid) using a JEOL JXA-8900 M electron microprobe with four wavelength-dispersive spectrometers. A selection of representative analyses has been included in Table 2, but the whole dataset is available as a supplementary electronic material (Supp. eTable 1). Analytical conditions were an accelerating voltage of 15 kV and an electron beam current of 20 nA, with a beam diameter of 5 µm. Elements were counted for 10 s on the peak and 5 s on each background position. Natural and synthetic standards were employed. Corrections were made using an on-line ZAF method. Detection limits are 0.02 wt% for Al, Na and K; 0.03 wt% for Ti, Fe, Mn, Mg, Ni and Cr, and 0.04 wt% for Si.

We have determined in situ concentrations of 28 trace elements (REE, Ba, Rb, Sr, Th, U, Nb, Ta, Pb, Zr, Hf, Y, Sc, V, Co and Cr) in olivine, plagioclase, orthopyroxene, clinopyroxene, amphibole and phlogopite on two polished thick sections (130 μ m) by laser ablation ICP–MS at the Natural History Museum of London, using a New Wave UP193FX

gas laser source operating at 193 nm coupled to an Agilent 7500CS ICP-MS. A selection of representative analyses has been included in Table 3, but the whole dataset is available as supplementary electronic material (Supp. eTable 2). Laser pulse energy was ca 0.2 mJ per pulse, energy density ca 3.5 J/cm², and frequency was set at 10 Hz. Laser ablated sample particles were carried in a flow of He gas to a high temperature (10 000 K) Ar plasma. The diameter of the laser beam was about 50 µm. Twenty analyses were performed in each run, the first two and last two corresponding to the calibration standard NIST 612 (a synthetic silicate glass; Pearce et al., 1997). The remaining 16 analyses were done on the selected samples. The counting time for one analysis was typically 90 s (40 s measuring gas blank to establish the background and 50 s for the remainder of the analysis). Analyses were monitored to detect accessory phase inclusions. Data were subsequently processed offline using LAMTRACE software, a programme written by Simon Jackson (Geological Survey of Canada) for handling time-resolved LA-ICP-MS data. Each analysis was normalized to Ca or Si using concentrations determined by electron microprobe. Detection limits for each element were in the range of 0.01-0.06 ppm, except for Sc and Cr (0.11 and 0.73 ppm, respectively) and a precision of 5% relative standard deviation was assumed.

4. Petrography

The studied samples represent medium- to coarse-grained olivine-bearing gabbronorites with intergranular texture, being olivine and plagioclase the early magmatic minerals (Fig. 2a, b). Olivine appears as anhedral, often rounded grains, whereas plagioclase forms euhedral to subhedral prisms with apparent concentric zoning.

The Talavera sample shows large anhedral poikilitic orthopyroxene crystals, leading to a hypidiomorphic subophitic

		oliv	vine		orthopyroxene					clinopyroxene										
	Talavera La Solanilla Talavera				vera	La So	lanilla			Talaver	a		La Solanilla							
Sample	107038	3L	10858	85L	107038L		1085	108585L			1070381	L		108585L						
Analysis	16c		08d	10c	04a	07c	05d	07d	015a	165	170	246	249	10	12	64	135	137		
SiO ₂	36.67	3	37.26	37.06	52.70	53.16	52.70	53.38	51.44	52.78	52.33	51.91	52.62	50.87	52.04	50.91	50.84	51.56		
TiO ₂	0.07	(0.01	0.02	0.28	0.39	0.20	0.06	0.88	0.61	0.61	0.49	0.24	0.73	0.31	0.45	1.09	0.91		
Al_2O_3	0.00	(0.01	0.25	1.18	2.06	1.29	1.07	2.41	2.72	2.51	2.55	1.49	3.23	1.39	3.14	5.47	4.50		
FeO	30.89	3	31.62	28.09	18.65	14.39	19.80	18.78	8.05	7.59	7.01	8.27	6.20	8.35	8.19	8.88	6.56	6.82		
NiO	0.04	(0.03	0.08	0.02	0.10	0.00	0.02	0.01	0.01	0.00	0.00	0.01	0.07	0.02	0.00	0.00	0.11		
MnO	0.01	(0.38	0.36	0.53	0.03	0.48	0.42	0.00	0.19	0.21	0.18	0.30	0.10	0.26	0.21	0.15	0.20		
MgO	33.52	3	31.78	33.26	24.97	26.90	23.65	25.43	16.51	17.40	15.35	16.71	15.14	16.46	17.18	18.90	14.80	15.22		
CaO	0.00	(0.02	0.33	0.85	2.53	0.46	0.25	19.54	17.63	20.41	18.15	22.66	18.63	19.12	15.73	19.06	18.85		
Na ₂ O	0.04	(0.00	0.09	0.04	0.06	0.00	0.00	0.40	0.29	0.44	0.48	0.43	0.71	0.35	0.40	1.12	0.87		
K_2O	0.02	(0.00	0.00	0.02	0.00	0.00	0.00	0.00	0.02	0.01	0.02	0.02	0.03	0.01	0.00	0.03	0.01		
Total	101.26	5 10	01.11	99.54	99.24	99.63	98.58	99.41	99.24	99.22	98.88	98.76	99.09	99.18	98.87	98.61	99.11	99.05		
Mg#	0.66	(0.64	0.68	0.70	0.77	0.68	0.71	0.79	0.80	0.80	0.78	0.81	0.78	0.79	0.79	0.80	0.80		
Si	0.982	1	.002	0.998	1.940	1.915	1.968	1.959	1.902	1.942	1.945	1.922	1.952	1.876	1.931	1.899	1.871	1.903		
Ti	0.001	0	0.000	0.000	0.008	0.011	0.006	0.002	0.025	0.017	0.017	0.014	0.007	0.020	0.009	0.013	0.030	0.025		
Al_{IV}	0.000	0	0.000	0.008	0.051	0.085	0.032	0.041	0.098	0.058	0.055	0.078	0.048	0.124	0.061	0.101	0.129	0.097		
Al_{VI}	-		-	-	0.000	0.003	0.025	0.005	0.007	0.060	0.055	0.033	0.017	0.016	0.000	0.032	0.108	0.098		
Fe	0.692	0).711	0.633	0.574	0.433	0.619	0.577	0.249	0.234	0.217	0.256	0.192	0.258	0.254	0.277	0.203	0.211		
Ni	0.001	0	0.001	0.002	0.001	0.003	0.000	0.001	0.001	0.001	0.000	0.000	0.001	0.002	0.001	0.000	0.000	0.003		
Mn	0.000	0).009	0.008	0.017	0.001	0.015	0.013	0.000	0.006	0.006	0.006	0.009	0.003	0.008	0.007	0.005	0.006		
Mg	1.339	1	.274	1.336	1.370	1.445	1.317	1.391	0.910	0.954	0.851	0.922	0.837	0.905	0.950	1.051	0.812	0.837		
Ca	0.000	0	0.001	0.009	0.033	0.098	0.018	0.010	0.774	0.695	0.813	0.720	0.901	0.736	0.760	0.629	0.752	0.745		
Na	0.002	0	0.000	0.005	0.003	0.004	0.000	0.000	0.028	0.021	0.032	0.035	0.031	0.051	0.025	0.029	0.080	0.062		
K	0.001	0	0.000	0.000	0.001	0.000	0.000	0.000	0.000	0.001	0.001	0.001	0.001	0.001	0.000	0.000	0.001	0.001		
Cations	3.018	2	2.998	2.999	3.998	3.998	4.000	3.999	3.994	3.989	3.992	3.987	3.996	3.992	3.999	4.038	3.991	3.988		
			I	Plagiocla	se					А	mphibo	le				Phlo	gopite			
Sample	10	070381	Ĺ		108	585L			1070)38L		1	1085851		1070)38L	108	585L		
Analysis	160	161	173	1	5	187	188	7b	8a	56	182	20	67	82	8b	34	10e	76		
SiO ₂	56.73 6	61.39	52.25	50.78	55.61	60.96	65.40	42.52	44.16	40.39	44.54	43.47	43.16	40.80	37.17	37.44	35.70	36.35		
TiO ₂	0.07	0.04	0.07	0.07	0.13	0.03	0.00	3.25	2.71	4.48	1.50	0.66	2.04	3.88	4.37	3.14	4.44	2.58		
Al_2O_3	27.71 2	24.30	30.77	30.78	27.57	24.75	21.12	11.75	10.49	12.19	11.59	13.85	12.23	11.92	15.37	16.08	14.38	18.02		
FeO	0.01	0.04	0.03	0.06	0.25	0.00	0.03	11.12	9.22	10.33	9.37	8.94	9.95	11.54	11.60	8.71	17.98	7.73		
NiO	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-		

7 mary 515	100	101	175	1	5	107	100	10	ou	50	102	20	07	01	00	51	100	10
SiO ₂	56.73	61.39	52.25	50.78	55.61	60.96	65.40	42.52	44.16	40.39	44.54	43.47	43.16	40.80	37.17	37.44	35.70	36.35
TiO ₂	0.07	0.04	0.07	0.07	0.13	0.03	0.00	3.25	2.71	4.48	1.50	0.66	2.04	3.88	4.37	3.14	4.44	2.58
Al_2O_3	27.71	24.30	30.77	30.78	27.57	24.75	21.12	11.75	10.49	12.19	11.59	13.85	12.23	11.92	15.37	16.08	14.38	18.02
FeO	0.01	0.04	0.03	0.06	0.25	0.00	0.03	11.12	9.22	10.33	9.37	8.94	9.95	11.54	11.60	8.71	17.98	7.73
NiO	-	-	-	-	-	-	-	_	-	-	-	-	-	-	-	-	-	-
MnO	0.02	0.01	0.03	0.00	0.06	0.00	0.03	0.02	0.07	0.13	0.11	0.16	0.15	0.09	0.00	0.00	0.18	0.00
MgO	0.00	0.00	0.00	0.00	0.00	0.00	0.05	13.64	15.30	13.59	14.98	15.38	14.79	12.84	16.73	19.39	12.81	20.03
CaO	9.46	5.94	12.83	13.96	9.33	6.18	2.38	11.18	11.25	11.33	10.91	11.30	11.00	11.40	0.02	0.01	0.13	0.06
Na_2O	6.39	8.47	4.05	4.07	6.67	8.41	10.77	2.45	2.48	2.37	2.70	2.65	2.62	2.54	0.64	0.94	0.21	1.22
K ₂ O	0.05	0.03	0.04	0.02	0.08	0.08	0.17	1.01	0.57	1.21	0.46	0.29	0.34	0.99	8.67	8.31	9.04	8.19
Total	100.5	100.3	100.1	99.8	99.7	100.5	100.0	97.0	96.3	96.1	96.2	96.7	96.3	96.1	94.6	94.0	94.9	94.2
Mg#	-	-	-	-	-	-	-	0.69	0.75	0.7	0.74	0.75	0.73	0.66	0.72	0.8	0.56	0.82
U																		
Si	10.14	10.88	9.46	9.28	10.06	10.80	11.54	6.57	6.78	6.06	6.67	6.54	6.36	6.15	5.48	5.48	5.45	5.29
Ti	0.009	0.005	0.010	0.010	0.018	0.004	0.000	0.378	0.313	0.506	0.169	0.074	0.226	0.440	0.485	0.345	0.511	0.282
Al _{IV}	5.834	5.072	6.557	6.625	5.871	5.163	4.388	1.429	1.219	1.940	1.333	1.464	1.638	1.854	2.518	2.524	2.546	2.712
Al_{VI}	-	-	-	-	-	-	-	0.708	0.679	0.216	0.709	0.988	0.486	0.262	0.155	0.247	0.044	0.377
Fe	0.001	0.006	0.005	0.009	0.038	0.000	0.004	1.437	1.184	1.296	1.172	1.124	1.227	1.454	1.431	1.065	2.297	0.940
Ni	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-
Mn	0.003	0.002	0.005	0.000	0.009	0.000	0.005	0.003	0.010	0.016	0.013	0.021	0.019	0.011	0.000	0.000	0.023	0.000
Mg	0.000	0.000	0.000	0.000	0.000	0.000	0.012	3.141	3.503	3.040	3.344	3.447	3.250	2.883	3.679	4.226	2.918	4.344
Ca	1.811	1.127	2.487	2.734	1.808	1.172	0.449	1.851	1.850	1.822	1.750	1.819	1.737	1.840	0.003	0.001	0.021	0.009
Na	2.214	2.912	1.422	1.440	2.337	2.890	3.685	0.735	0.740	0.689	0.784	0.772	0.749	0.742	0.183	0.266	0.063	0.345
Κ	0.012	0.007	0.009	0.005	0.018	0.017	0.039	0.199	0.112	0.232	0.088	0.055	0.063	0.191	1.632	1.550	1.761	1.521
Cations	20.03	20.02	19.95	20.10	20.16	20.05	20.12	16.45	16.39	15.82	16.03	16.30	15.76	15.82	15.57	15.70	15.64	15.82

Table 2.- Major element composition (wt.%) of representative minerals from SCS gabbros

	C	olivine			clinopyroxene												
	Talavera La Solanilla T				ivera	La So	lanilla			Falavera	a		La Solanilla				
Sample	107038L 108585L			1070)38L	108	108585L		1	070381	L		108585L				
Analysis	16c	08d	10c	04a	07c	05d	07d	015a	166	171	247	250	11	13	65	136	138
Ва	0.316	0.068	0.054	0.134	0.079	0.065	1.25	14.8	0.110	9.69	5.31	6.63	6.93	16.7	1.19	5.14	1.54
Rb	0.068	0.102	0.058	0.061	0.056	0.055	0.371	1.54	0.040	1.49	1.00	6.81	0.150	0.300	0.037	0.610	0.390
Sr	0.197	0.130	0.421	0.454	0.329	0.229	0.118	27.4	14.7	20.9	18.4	19.4	20.3	62.5	15.1	29.8	25.2
Pb	0.080	0.046	0.051	0.069	0.056	0.037	0.052	0.658	0.400	0.680	0.870	1.860	0.410	0.770	0.046	0.340	0.410
Th	0.019	0.055	0.025	0.022	0.304	0.397	0.730	0.842	0.310	1.06	1.01	0.810	0.610	0.730	0.102	0.160	0.440
U	0.014	0.019	0.010	0.015	0.037	0.059	0.166	0.102	0.090	0.170	0.140	0.130	0.050	0.050	0.013	0.020	0.040
Zr	0.716	0.705	0.531	10.1	8.65	30.0	11.2	81.3	24.6	59.4	72.9	99.8	38.4	64.5	21.6	48.7	56.9
Nb	0.044	0.022	0.015	0.036	0.022	0.156	0.100	0.447	0.050	0.530	0.810	0.170	0.060	0.320	0.030	0.200	0.050
Y	0.094	0.264	0.269	13.2	8.03	11.1	3.56	63.0	24.0	37.7	38.6	42.6	34.5	37.5	23.8	35.9	42.0
Sc	6.63	3.08	4.77	29.1	75.6	57.5	6.06	181	103	115	94.1	93.6	122	127	120	145	141
Со	221	207	184	71.8	79.1	76.9	70.8	30.7	43.4	44.1	37.4	41.8	46.1	50.8	34.7	41.2	41.5
V	2.68	3.00	3.79	65.6	265	91.6	52.7	540	607	657	496	276	568	545	333	637	656
Cr	3.48	189	134	68.5	766	5.95	4.07	751	5710	6090	4500	1050	2100	533	2130	2860	3180
Та	0.003	0.013	0.018	0.019	0.007	0.017	0.015	0.042	0.010	0.030	0.060	0.010	0.010	0.010	0.005	0.020	0.010
Hf	0.089	0.107	0.092	0.409	0.322	1.27	0.824	3.44	1.11	1.54	2.00	3.70	1.34	2.20	0.823	2.12	2.17
La	0.016	0.049	0.035	0.150	0.193	0.468	0.132	6.92	1.91	7.14	6.85	6.94	2.90	3.94	1.06	1.70	2.38
Ce	0.018	0.128	0.046	0.354	0.498	0.918	0.254	16.7	7.40	26.6	24.1	26.2	8.99	12.3	3.24	6.32	8.35
Pr	0.012	0.029	0.011	0.094	0.068	0.179	0.040	3.53	1.36	4.15	4.19	4.64	1.61	2.07	0.714	1.38	1.56
Nd	0.176	0.153	0.242	0.849	0.568	1.08	0.405	22.5	8.45	21.4	21.5	22.9	8.55	12.7	4.74	8.32	9.64
Sm	0.063	0.073	0.013	0.437	0.321	0.692	0.145	8.10	2.98	5.58	5.94	6.66	3.15	4.22	2.03	3.46	3.93
Eu	0.021	0.020	0.026	0.052	0.052	0.073	0.027	1.46	0.740	1.120	1.100	0.920	0.720	0.830	0.600	1.01	1.21
Gd	0.164	0.077	0.062	0.812	0.516	0.912	0.259	10.6	3.84	6.01	6.66	7.48	4.89	5.44	3.37	5.12	5.75
Tb	0.012	0.015	0.011	0.222	0.127	0.202	0.051	1.71	0.670	1.00	1.17	1.24	0.900	0.960	0.571	0.960	1.05
Dy	0.030	0.048	0.048	1.69	1.10	1.91	0.489	12.4	4.77	6.89	7.70	8.09	6.51	7.30	4.42	6.76	7.56
Но	0.010	0.014	0.014	0.410	0.295	0.419	0.144	2.53	0.920	1.48	1.56	1.61	1.33	1.49	0.930	1.44	1.64
Er	0.061	0.039	0.055	1.69	1.19	1.54	0.462	6.55	2.73	4.07	4.46	4.53	4.08	4.12	2.76	4.24	4.89
Tm	0.014	0.022	0.014	0.373	0.231	0.293	0.076	0.955	0.370	0.600	0.570	0.650	0.600	0.590	0.414	0.570	0.690
Yb	0.109	0.055	0.095	3.10	1.75	2.83	0.726	5.67	2.32	4.24	3.79	4.54	3.98	4.44	2.78	4.21	4.55
Lu	0.044	0.019	0.013	0.575	0.260	0.548	0.127	0.838	0.330	0.580	0.560	0.630	0.580	0.550	0.409	0.590	0.650

				A	mphib	Phlogopite												
Sample		107038	L		108585L				1070)38L			108585	L	1070)38L	1085	85L
Analysis	160	161	173	1	5	187	188	7b	8a	56	182	20	67	82	8b	34	10e	76
Ва	82.6	93.6	88.3	65.7	136	280	209	239	17.4	33.7	171	54.4	91.6	295	3140	1250	976	3340
Rb	0.130	0.240	0.058	0.780	2.68	1.42	0.16	5.13	1.53	2.87	4.42	0.97	1.7	2.89	374	326	346	288
Sr	538	447	606	706	729	822	369	92.6	97.6	51.3	89.9	48.0	50.6	95.8	9.07	19.8	13.9	31.4
Pb	12.1	11.3	6.59	2.46	6.29	7.63	3.72	2.33	4.3	1.58	2.82	1.25	1.02	1.91	4.91	5.48	2.41	4.36
Th	0.015	0.017	0.010	0.030	0.014	0.050	0.016	2.06	3.98	2.22	2.03	8.59	0.92	0.92	0.02	0.01	0.01	0.02
U	0.012	0.011	0.007	0.010	0.005	0.050	0.010	0.58	0.24	0.29	0.36	1.25	0.086	0.055	0.01	0.01	0.00	0.01
Zr	0.038	0.076	0.040	1.14	0.170	0.220	0.038	819	521	1240	230	250	430	603	25.3	12.4	3.84	2.11
Nb	0.041	0.054	0.014	0.033	0.021	0.140	0.012	37.7	21.5	45.4	24.6	13.4	36.9	27.4	36.7	34.6	13.1	15.0
Y	0.390	0.080	0.560	0.470	0.320	0.540	0.090	206	209	183	123	223	300	400	0.05	0.02	0.28	0.05
Sc	2.74	3.00	2.77	4.11	4.06	3.73	3.11	123	152	93.0	95.8	14.3	59.0	212	15.8	11.3	13.1	8.49
Co	0.051	0.085	0.080	0.310	0.640	0.140	0.074	43.9	48.7	46.5	57.1	46.9	43.8	48.4	92.2	93.0	93.2	67.2
V	0.330	0.280	2.44	2.47	3.87	0.200	0.090	438	922	334	609	95.8	440	514	435	435	710	715
Cr	1.83	1.71	1.03	1.49	1.45	1.27	1.55	688	1600	486	565	2.9	4.66	17.6	1370	354	43.0	128
Та	0.016	0.019	0.001	0.012	0.012	0.040	0.018	2.67	1.43	3.91	1.36	0.8	3.3	1.81	2.45	2.70	1.20	1.27
Hf	0.033	0.093	0.027	0.059	0.094	0.076	0.062	24.8	22.5	42.0	7.86	9.61	23.8	22.0	1.18	0.563	0.177	0.050
La	16.9	9.07	2.96	1.71	1.56	6.95	7.71	44.3	28.8	47.8	26.5	75.7	37.9	47.3	0.015	0.013	0.256	0.029
Ce	20.6	11.1	5.29	3.13	2.80	8.43	7.86	120	80.2	140	93.3	216	122	138	0.011	0.010	0.401	0.033
Pr	1.52	0.710	0.570	0.310	0.290	0.650	0.510	23.0	13.3	26.1	15.8	30.6	24.4	27.7	0.007	0.006	0.040	0.009
Nd	4.45	2.76	1.99	1.01	1.29	2.06	1.41	126	75.8	128	82.2	144	129	157	0.079	0.082	0.231	0.136
Sm	0.510	0.097	0.340	0.110	0.110	0.220	0.140	34.7	26.9	34.1	24.1	35.3	40.3	50.6	0.053	0.079	0.061	0.070
Eu	1.67	1.34	0.970	0.670	1.27	3.10	2.03	4.23	3.71	3.62	3.82	5.46	4.04	10.1	0.034	0.062	0.079	0.193
Gd	0.180	0.190	0.250	0.076	0.110	0.130	0.090	37.7	32.5	35.0	25.9	38.1	44.2	63.2	0.048	0.113	0.050	0.075
Tb	0.013	0.020	0.030	0.019	0.010	0.013	0.016	5.74	5.70	5.41	3.91	5.99	7.69	11.0	0.011	0.008	0.009	0.010
Dy	0.100	0.053	0.100	0.130	0.090	0.080	0.040	39.1	39.3	35.5	25.3	40.4	54.6	76.6	0.070	0.031	0.043	0.051
Но	0.015	0.018	0.020	0.020	0.010	0.020	0.010	7.99	8.06	7.09	4.96	8.37	11.5	16.0	0.005	0.006	0.012	0.006
Er	0.040	0.048	0.100	0.090	0.030	0.061	0.061	21.9	23.6	18.7	13.3	23.9	34.4	45.8	0.047	0.049	0.028	0.028
Tm	0.007	0.013	0.008	0.010	0.016	0.018	0.006	2.96	3.23	2.60	1.77	3.40	5.18	6.71	0.011	0.008	0.005	0.008
Yb	0.058	0.078	0.105	0.079	0.060	0.080	0.076	19.5	21.0	16.5	11.1	22.0	33.0	42.6	0.038	0.056	0.052	0.058
Lu	0.013	0.015	0.011	0.010	0.016	0.020	0.008	2.77	2.95	2.19	1.49	3.01	4.66	6.07	0.011	0.006	0.009	0.006

Table 3.- Trace element composition (ppm) of representative minerals from SCS gabbros

to ophitic texture. Interstitial minerals are clinopyroxene, amphibole, phlogopite and ilmenite, together with Na-rich plagioclase overgrowths (Fig. 2a, b). Amphibole and phlogopite are relatively abundant (~9%) and exhibit a relatively large size (Fig. 2c, d) (similar to that of clinopyroxene; i.e. ~ 1–3 mm). A pale green amphibole is also present in much lower proportions, generally forming thin margins around large brown amphiboles (Fig. 2d).

Early magmatic minerals from La Solanilla sample (olivine and plagioclase) constitute more than 70% of the rock. Oli-

vine from this gabbro is characterised by abundant Cr-spinel inclusions and development of complex coronas formed by an inner shell of orthopyroxene (sometimes consisting of radial columnar grains) and a second shell of Ti-poor pargasitic amphibole (Fig. 2e). Orthopyroxene only appears in these thin coronas. The abundance of amphibole and phlogopite (~6%) is lower than in the Talavera sample, as well as their size (< 1 mm). While pale green amphibole is restricted to coronas around olivine, brown amphibole appears within interstitial pockets throughout the sample, associated with phlogopite



Fig. 2.- Plane polarized light (a–e) and back scattered electron (f) images showing the main textures of Talavera and La Solanilla gabbroic intrusions. a–b: General texture of Talavera and La Solanilla gabbros, respectively. c: Interstitial phlogopite with zircon and apatite inclusions (Talavera sample). d: Clinopyroxene being replaced by amphibole in Talavera sample. Note that two types of amphibole are present: a large brown amphibole and a smaller pale green amphibole (center and right respectively). e: Composite orthopyroxene–pale-green amphibole corona in olivine–plagioclase contact (La Solanilla sample). f: Back scattered electron image showing the disequilibrium texture between clinopyroxene and amphibole (Talavera sample). A detail of the reaction zone is expanded (circle), where small amphibole inclusions growing within the clinopyroxene crystal are apparent. White bar in images a–e represent 250 μm.

and ilmenite (Fig. 2b). Subsolidus alteration is manifested by the formation of a secondary paragenesis mainly consisting of chlorite, sericite and a secondary green- to colourless amphibole (Mg-hornblende to actinolite and cummingtonite) at the expense of the primary minerals.

A main characteristic of both samples (although it is more notorious in the Talavera gabbro) is the common presence of reaction textures around clinopyroxene, which is variably replaced by amphibole. This is also manifested by the growth of small oriented amphibole inclusions ($2 \times 10 \ \mu m$) within clinopyroxene rims (Fig. 2f).

The main accessory minerals of these gabbros appear within the interstitial matrix (Fig. 2c), and are represented by ilmenite (which is always included in amphibole and phlogopite), apatite, zircon, baddeleyite and titanite. Zircon crystals are more abundant in the Talavera gabbro when compared with La Solanilla (see Villaseca *et al.*, 2011 for further details). This mineral can be found either as isolated large grains or as second-generation lamellar crystals at the ilmenite rims.

5. Mineral composition

5.1. Major elements

Major element composition of the main minerals is heterogeneous in both gabbroic intrusions (Table 2, Supp. eTable1 and Fig. 3). The mafic phases are good examples of this heterogeneity: the forsterite content of olivine ranges from Fo_{78} to Fo₆₃ (see Orejana et al., 2009), and the Mg# values of orthopyroxene and clinopyroxene are 0.82–0.62 and 0.84–0.69, respectively (Fig. 3a,b; Table 2). Orthopyroxene composition displays a marked contrast when comparing the two gabbroic intrusions, with Talavera samples yielding higher Mg# (0.82–0.7) and CaO (0.85–2.5 wt.%) than coronitic orthopyroxene in the La Solanilla samples (Fig. 3a). Clinopyroxene exhibits the following major element values: CaO (15-22.7 wt.%), Al₂O₂ (1.3-7.4 wt.%), Na₂O (0.3-1.78 wt.%) and TiO₂ (0.1-1.5 wt.%), with positive correlation between Al, Ti and Na (Fig. 3b, c, d). Slight chemical differences can also be observed between clinopyroxene from the Talavera and the La Solanilla gabbros, with the former showing the highest CaO, and lowest Na₂O and Al₂O₂ contents (Fig. 3b ,c ,d). Clinopyroxene analyses showing the highest Al₂O₂, TiO₂ and Na₂O values in the La Solanilla gabbro (Fig. 3c, d) represent areas partially transformed to amphibole.

Plagioclase is characterised by a strong chemical variation, with anorthite values from An_{s2} to An_{10} , $SiO_2 = 42-66$ wt.% and $Al_2O_3 = 21.1-34$ wt.% (Fig. 3e, f). Amphibole has a heterogeneous composition in Mg# (0.62–0.78), TiO_2 (0–4.7 wt.%), Al_2O_3 (10.4–16 wt.%), Na_2O (1.8–3.05 wt.%) and K_2O (0.2–1.15 wt.%) (Fig. 3g, h, i), and displays positive correlations for Ti–K and Na–Al, and negative correlation for Ti–Mg# (Fig. 3g, h, i). It can be classified as pargasite, magnesiohastingsite, magnesiohornblende and kaersutite (following Leake *et al.*, 1997 criteria). Phlogopite also shows a significant within-sample variation in Mg# (0.56–0.86), TiO_2 (0.32–4.7 wt.%), Al_2O_3 (14.4–18 wt.%), Na_2O (0.2–1.3 wt.%) and K_2O (7.4–9.1 wt.%) (Fig. 3j, k, l). It also exhibits negative correlation between Ti and Mg# and Al (Fig. 3j, k), so that a few phlogopite crystals display very low TiO₂ values (down to 0.32 wt.%) with Mg# higher than 0.78 (Fig. 3j).

Some of the main minerals present chemical zoning. In clinopyroxene crystals, the TiO₂, Al₂O₃ and Na₂O concentration decreases from core to rim (Fig. 4). The chemical heterogeneity of plagioclase is also manifested in marked core to rim zoning patterns within a single crystal (e.g., anorthite from An₆₁ to An₁₀). The most An-depleted plagioclase corresponds to rims and anhedral interstitial crystals adjacent to subhedral plagioclase prisms (Fig. 4). Analysed amphibole crystals do not display important zoning patterns, but they reveal that two broad compositional poles can be distinguished regarding their major element contents (with a few intermediate analyses): 1) large brown amphibole with high TiO₂ and K₂O, and 2) brown to green amphibole with lower concentrations of these elements (Fig. 3).

5.2. Trace elements

The complete set of trace element data is available in Supp. eTable 2. Olivine trace element contents are characterised by moderate to low values for the transition metals: Ni (212–1570 ppm), Cr (up to 190 ppm), Co (184–242 ppm), V (0.6–3.8 ppm) and Sc (3–7.7 ppm) (Table 2 and 3). Olivine from the Talavera sample displays higher Sc and Ni values, and lower Cr and V, with respect to those from La Solanilla gabbro (Tables 1 and 2).

The large locally poikilitic orthopyroxene crystals of Talavera sample (107038L) yield a variable compositional range with high Cr (68–4410 ppm) contents, which correlate positively with Mg# (Fig. 6a). Coronitic orthopyroxene in the La Solanilla sample (108585L) is characterised by very low Cr (4–14 ppm), but higher concentrations of many incompatible elements: Ba (up to 1.5 ppm), Rb (up to 0.4 ppm), Th (0.4–0.73 ppm), Zr (11–30 ppm), Nb (up to 0.16 ppm) and LREE (e.g. Ce up to 0.92 ppm), when compared with Talavera orthopyroxene (Fig. 6a, b, c). Primitive mantle-normalised plots of trace element abundances of orthopyroxene display a consistent fractionation in the REE towards lower LREE and characteristic positive Th–U peaks and Sr–Eu negative anomalies (Fig. 6d).

The main differences in plagioclase trace element contents between the two samples are the higher Rb and Sr, and lower LREE in La Solanilla with respect to Talavera gabbro (Table 3; Fig. 6e, f, g). In both cases there is a general trend of decreasing Sr and increasing Ba and LREE associated with higher Na₂O contents (Fig. 6f, g). This variation is mostly



Fig. 3.- Major element composition (wt.%) of (a) orthopyroxene, (b–d) clinopyroxene, (e–f) plagioclase, (g–i) amphibole and (j–l) phlogopite from the Talavera and La Solanilla gabbros. Clinopyroxene analyses corresponding to mineral areas partially transformed to amphibole have been indicated in diagrams c and d. Part of the orthopyroxene data plotted in diagram a have been extracted from Orejana *et al.* (2009).

mimicked by core to rim trace element zoning within a single euhedral crystal, but a significant drop in Ba and LREE is characteristic of extremely Na-rich plagioclase overgrowths (Fig. 4). Plagioclase also shows fractionated LREE chondrite-normalised patterns (La up to 80x chondrite abundances; Fig. 6g).

Clinopyroxene from Talavera shows higher Cr, Th and U contents (e.g., Cr up to 6170 ppm, Th up to 2.86 ppm and U

up to 0.3 ppm; Fig. 7a, b) with respect to La Solanilla (Table 3). LREE and HFSE (high field strength elements) do not differ significantly between both intrusions (Fig. 7c, d), but HREE (heavy REE) are higher in cpx of the La Solanilla gabbro (Fig. 7e). Most incompatible trace elements in La Solanilla clinopyroxene (Ba, Th, Pb, Zr, Hf, Y, REE) are negatively correlated when using Cr content as a differentiation index (see examples in Fig. 7a, b, c, d). This general trend is similar



Fig. 4.- Major (wt.%) and trace (ppm) element zoning profiles within single clinopyroxene (left) and plagioclase (right) crystals from Talavera and La Solanilla samples. Back scattered electron images at the top of each column illustrate the location of the analysis spots (white circles). Clinopyroxene core analyses in the plots are shown with a grey band. Trace element concentrations in plagioclase have been determined only in the spots located within the white dotted rectangle. Note that analyses from La Solanilla sample have been performed in several plagioclase prisms, thus causing the observed irregular compositional patterns.

to that depicted by core to rim profiles, and it is interesting to remark that analyses with the highest incompatible trace elements values correspond with partially altered clinopyroxene (areas replaced by amphibole). On the contrary, clinopyroxene from Talavera is characterised by: 1) a fairly variable trace element concentration in analyses displaying very high Cr contents, 2) a slight tendency of Ba, Th, U, Pb, Zr, Nb, V and LREE contents to decrease towards lower Cr, and 3) the presence of a reduced group of spots with low Cr values but relative high incompatible trace element concentrations (Ba, U, Th, Pb, Zr, Hf, Y, REE), as described in the La Solanilla gabbros (Fig. 7a, b, c, d). These latter analyses represent partially transformed zones in contact with interstitial amphiboles.

Clinopyroxene chondrite-normalised REE patterns are almost flat, but exhibit a wide compositional range and a variable degree of LREE fractionation (Fig. 7e). They always show Eu negative anomaly. The Primitive Mantle-normalised patterns display characteristic Nb–Ta, Sr and Ti negative anomalies (Fig. 7f). Zoning patterns in both samples are in agreement with development of rims in reaction fronts (and also inner areas) displaying lower Cr and higher values for incompatible elements (Fig. 4). It must be noted that, excluding partially transformed clinopyroxene areas, the extreme heterogeneity of incompatible elements in Talavera sample, such as Th, U, Pb, Zr, Nb or LREE, corresponds with analyses displaying the highest Cr concentrations (> 5000 ppm) (Fig. 7a, b, c, d).

Cr content in amphibole from Talavera (249-796 ppm) is again significantly higher than in La Solanilla gabbro (< 96 ppm) (Table 3). The trace element composition of amphibole from both intrusions shows a broad tendency of the low Ti-K group to show lower Ba, Rb, Sr and V concentrations when compared with the Ti-K rich amphiboles. An example of this behaviour can be observed in the zoning profile of Figure 5, where two distinct types of amphiboles (brown and green) have been analysed, with green amphibole being relatively depleted in Ti, K, Sr, Ba and LREE. HFSE and REE also decrease their concentration in amphiboles with lower Ti (Fig. 8a, b, c, d). The amphibole associated with coronitic orthopyroxene does not show a distinct composition with respect to other interstitial amphiboles from La Solanilla. The chondrite-normalised trace element values of amphiboles are characterised by very high REE abundances and negative anomalies at Eu, and the primitive mantle-normalised patterns display negative anomalies at Pb, Sr and Ti (Fig. 8e, f).

Similarly to other mafic minerals, there is a contrasted trace element composition in phlogopite from both samples, manifested in the high and variable Cr contents of Talavera (354–1370 ppm), when compared to La Solanilla (3.4–152 ppm) (Table 3). Moreover, phlogopite from this latter gabbro reaches higher Ba (up to 6430 ppm), Y (up to 0.28 ppm) and LREE (e.g. Ce up to 0.4 ppm) values, and lower Pb and HFSE contents (e.g., Fig. 9a, b). It is noteworthy the heterogeneity in the phlogopite Zr–Hf contents from Talavera (Fig. 9b). Primitive Mantle-normalised patterns of phlogopite are characterised by strong abundances of Rb–Ba, Nb–Ta, Pb and Ti, and negative Th–U, LREE and Y anomalies (Fig. 9c).

6. Discussion

6.1. Constraints on processes during magma crystallization

With the exception of olivine, all mafic phases present in the La Solanilla sample are characteristically depleted in Cr with respect to the same minerals in the Talavera gabbro, irrespective of their Mg# (Table 3; Fig. 6a; Fig. 7a, b, c, d). This fact might be explained by the much higher abundance of Cr-spinel inclusions in olivine from La Solanilla (Fig. 2e), whose crystallization depleted drastically the Cr content in the residual melt.

Unlike other main minerals, orthopyroxene displays contrasting textural and chemical characteristics in the Talavera and La Solanilla samples. Orthopyroxene from Talavera gabbro yields a negative correlation between incompatible elements and Mg# (e.g., Fig. 6c), coherent with a progressive enrichment of the residual liquid in those elements. On the

Fig. 5.- (a-c) Major and trace element zoning profiles within a single amphibole crystal from Talavera sample. The back scattered electron image (a) illustrates the location of the analysis spots (white circles). Major and trace element contents are wt% and ppm, respectively. Th and Sr concentrations are x100 and x2, respectively. Note that the image (a) corresponds with Fig. 2d









Fig. 7.- Trace element composition (ppm) (a–d) and Chondrite- or Primitive Mantle-normalised patterns (e–f) of clinopyroxene from Talavera and La Solanilla gabbros. In diagrams a, b, c and d, the solid black arrow represents the compositional variation trend in Talavera sample, the wide grey double arrow indicates the compositional range for Cr-rich analyses in Talavera gabbro, and the grey filled symbols represent areas partially replaced by amphibole. Note that trace elements in diagrams a, b, c and d have been plotted against a parameter illustrative of the degree of differentiation. Chondrite and Primitive Mantle normalising values after Sun and McDonough (1989) and McDonough and Sun (1995), respectively.



Fig. 8.- Trace element composition (ppm) (a–d) and Chondrite- or Primitive Mantle-normalised patterns (e–f) of amphibole from Talavera and La Solanilla gabbros. Note that trace elements in diagrams a, b, c and d have been plotted against a parameter illustrative of the degree of differentiation. REE in plot c represents the sum of La to Lu concentrations (ppm). Chondrite and Primitive Mantle normalising values after Sun and McDonough (1989) and McDonough and Sun (1995), respectively.

other hand, orthopyroxene from La Solanilla, which only appears forming coronas around olivine, have lower Cr contents and Mg# values (always bellow 0.74; Fig. 6a) and much higher concentrations for certain incompatible trace elements

(e.g., Fig. 6b, c). The presence of composite orthopyroxene– amphibole coronas at olivine–plagioclase contacts (Fig. 2e) has been previously described and attributed to a magmatic origin, rather than metamorphic (e.g. Joesten, 1986; Haas *et* *al.*, 2002). The relatively high concentration of incompatible trace elements (such as LILE and REE) in the coronitic orthopyroxene from La Solanilla (Fig. 6b, c, d) does not favour formation by subsolidus processes, but rather require the involvement of an evolved interstitial silicate liquid in its origin, thus reinforcing the possibility of a late magmatic genesis.

The heterogeneous chemical composition shown by the early-crystallisation phases of the SCS gabbros has been related to some degree of crystal fractionation (Orejana *et al.*, 2009). The continuous normal zoning displayed by plagioclase in the two analysed samples for the present study, characterised by Ab-rich contents towards the rim of idiomorphic prisms (Fig. 4), agrees with the above conclusion. Widespread plagioclase crystallization is also consistent with the negative Sr and Eu anomalies in the normalised patterns of orthopyroxene, clinopyroxene and amphibole (Fig. 6d; Fig. 7e; Fig. 8e).

On the other hand, chemical zoning in clinopyroxene, either in major or trace elements, is likely influenced by element diffusion during its replacement by amphibole (Fig. 2d. f), which can be observed in both samples (Talavera and La Solanilla). The decrease in Ti, Al and Na (Fig. 4) would be in accordance with their preferential diffusion towards crystallising amphibole, which have higher partition coefficients for these elements. Similar textural and chemical features in clinopyroxene-amphibole reaction areas have been identified in oceanic gabbros by Coogan et al. (2001). Clinopyroxene analyses with the highest Al, Ti and Na concentrations (and lower Ca) in La Solanilla gabbro (Fig. 3c, d) usually display a low total cation sum (~97-98 wt%), likely indicating that they represent secondary highly transformed mineral domains. The fact that clinopyroxene rims are enriched in trace elements more compatible with amphibole (Fig. 4) is likely the result of analysing with a near 50 µm diameter laser beam, which has likely ablated part of the $< 10 \ \mu m$ amphibole micro-inclusions present in clinopyroxene rims associated with reaction zones (Fig. 2f). Due to the uncertainty in the composition of analyses from partially transformed areas (which have been marked in Fig. 3c, d and Fig. 7a, b, c, d), they have not been considered in further discussions.

The above features reveal a complex history of interstitial melt in-situ crystallization. Several textural and chemical observations testify for this process both in the Talavera and La Solanilla gabbroic intrusions. The evidences are: 1) the presence of interstitial minerals rich in highly incompatible trace elements (mainly amphibole, phlogopite, ilmenite, apatite, zircon, baddeleyite and titanite), 2) overgrowth of early crystallization phases towards a more differentiated composition (mainly plagioclase), and 3) disequilibrium between some interstitial crystals and previous minerals (e.g. clinopyrox-ene–amphibole reaction). Such features have been commonly interpreted as resulting by late-stage crystallization of interstitial melts (e.g. Barnes, 1986; Bédard, 1994; Ross and El-thon, 1997; Tribuzio *et al.*, 1999b). Complete re-equilibration of the early-formed minerals with the residual melt fraction



Fig. 9.- Trace element composition (ppm) (a–b) and Primitive Mantle-normalised patterns (c) of phlogopite from Talavera and La Solanilla gabbros. Note that trace elements in diagrams a and b have been plotted against a parameter illustrative of the degree of differentiation. Primitive Mantle normalising values after McDonough and Sun (1995).

was not attained, as demonstrated by the late crystallization of interstitial plagioclase with anorthite contents occasionally below 10, together with the preservation of chemical zoning in this mineral (Fig. 4). The similarity in Mg# values between the early- and late-crystallized mafic minerals (Mg#_{ol} = 0.78– 0.63, Mg#_{Opx} = 0.82–0.68, Mg#_{Cpx} = 0.84–0.74, Mg#_{Amph} = 0.75–0.64, Mg#_{phl} = 0.82–0.56), results somewhat contradictory. However, this feature could be explained in two ways: 1) due to a significant decrease of Fe in the residual melt (see below the discussion regarding the influence of ilmenite), or 2) advocating a significant increase in the oxygen fugacity, which can induce higher Mg# values in hydrous minerals (Scaillet and Evans, 1999). The increment of oxygen fugacity likely occurred in conjunction with the increase of the H₂O concentration in the late-stage residual melt.

The formation of Ti-rich amphibole in gabbroic rocks has been ascribed in the literature to a metamorphic or magmatic origin. The first possibility has been mainly proposed for solid-state reactions during infiltration of seawater-derived

7000

5000

3000

1000

30

20

й

(b)

Ba

(a)

Talavera

□La Solanilla



Fig. 10.- (a) Na_2O vs. $(La/Sm)_N$ values in plagioclase and (b) La vs. $(La/Sm)_N$ in clinopyroxene from Talavera and La Solanilla gabbros. Only clinopyroxenes with the highest Cr contents have been plotted in diagram (b). Analyses affected by transformation of clinopyroxene have not been considered. Solid and dotted black arrows represent compositional variation trends. La and Sm concentrations normalised to Chondrite contents (Sun and McDonough, 1989)

fluids through mafic plutonic rocks in oceanic environments (e.g. Coogan et al., 2001 and references therein). Sub-solidus hydrothermal alteration is highly unlikely in the SCS gabbroic bodies considering the temperature estimates (825 to 970 °C, hornblende-plagioclase geothermometer; Molina et al., 2009). The very high REE, Y, Zr and Ti concentrations of amphibole from Talavera and La Solanilla intrusions (Fig. 8a, e, f), and the fact that amphibole is texturally in equilibrium with other magmatic interstitial phases (e.g. phlogopite, apatite), strongly support its magmatic genesis in the studied samples. The occurrence of Ti-rich amphibole as an interstitial phase in gabbroic rocks, associated with a priori anhydrous incompatible trace element-depleted systems, have been widely explained by late crystallization from more fractionated magmas, after liquidus minerals formation (mostly in tholeiitic basic magmas; e.g. Gillis, 1996; Tribuzio et al., 1999b, 2000; Coogan et al., 2001).

The heterogeneity of clinopyroxene composition in both gabbros might have fairly different origins. As already men-

tioned, some analyses from both samples are affected by the replacement of clinopyroxene by amphibole (e.g. see Fig. 3c, d and Fig. 7a, b, c, d), and they are not considered for further discussions. However, a subtle increment in trace element contents towards lower Cr can be observed in clinopyroxene from the La Solanilla gabbro (Fig. 7a, b, c, d), which points to a progressive enrichment of the residual melt in incompatible elements. This is also suggested by the LREE-enriched nature of interstitial An-poor plagioclase (Fig. 10a). On the contrary, clinopyroxene from the Talavera sample has two contrasting features: 1) extremely variable trace element contents for clinopyroxene with high Cr values, and 2) decreasing Ba, Th, U, Zr, Nb, V and LREE towards lower Cr concentrations (e.g., Fig. 7a, b, c, d). While the first characteristic will be discussed in more detail in the following section, the depletion of some trace elements in the most evolved clinopyroxene might be interpreted as due to co-precipitation of phases with high partition coefficients for certain elements, such as apatite (LREE), zircon (Zr, Hf, Th, U), baddeleyite (Zr, Hf, U), amphibole (Nb) and phlogopite (Ba, Nb, V). The influence of co-crystallising minerals in Talavera can also be identified in the amphibole chemistry. The large brown Ti-K-rich kaersutite likely represents the initial amphibole composition, whereas the lower Ti, K, Rb, Ba, Sr and V contents of the more interstitial green pargasite (Fig. 5) would be caused by coeval crystallization of phlogopite and Ab-rich plagioclase (overgrowths). Nevertheless, the fact that low Ti-K amphibole from Talavera shows an heterogeneous composition for Zr-Hf, Nb-Ta, Th and LREE (e.g. Zr-Nb, Fig. 8a, b), implies that amphibole formation in equilibrium with zircon, baddelevite, apatite and ilmenite was a local phenomenon. Thus, while phlogopite and interstitial plagioclase seem to have controlled the distribution of LILE and Sr, respectively, in the residual melt, the local precipitation of the above accessory phases have governed HFSE and LREE distribution. More specifically, the relevant role of ilmenite can be also identified in the behavior of Mg# and Ti in amphibole and phlogopite. Both minerals are in equilibrium with ilmenite and display moderate to low TiO, associated to high Mg#, which is an effect that could be expected from the crystallization of a Ti-Fe-rich mineral.

To evaluate more precisely the role played by accessory phases in the trace element distribution we have made a mass balance calculation considering the observed modal proportions of olivine, plagioclase, orthopyroxene, clinopyroxene, amphibole and phlogopite. Several conclusions for both samples can be drawn from the results (see calculation details in caption to Fig. 11): 1) the distribution of most trace elements is controlled by minerals crystallised from the residual interstitial liquid (amphibole, phlogopite and other accessory phases), excepting the HREE, that are significantly incorporated in pyroxene (\sim 30–50%), and Ba and LREE in plagio-clase (\sim 10–30%); 2) HREE are largely controlled by amphibole, 3) LILE are mainly incorporated into phlogopite (Rb, Ba) and plagioclase (Ba), 4) Nb–Ta demonstrate a clear pref-

erence for amphibole, phlogopite and one accessory mineral (likely ilmenite), and 5) many incompatible trace elements seem to be predominantly controlled by accessory phases present in modal proportion below 1 %: LREE (apatite) and Th–U and Zr–Hf (zircon and baddeleyite) (Fig. 11).

6.2. Involvement of an external component

Although textural and chemical criteria support the precipitation of interstitial minerals from a residual liquid by in-situ crystallization, this process might have progressed either in a closed or in an open system situation.

To discuss these possibilities we have applied the Equilibrium Distribution Method (EDM) of Bédard (1994). This author proposed a mass balance calculation to model the distribution of trace elements between a mineral assemblage and an interstitial trapped melt during crystallization in a closed system, applicable to cumulate rocks or any other solid-liquid system. The equation of Bédard (1994) allows the calculation of any mineral composition if the following variables are known: 1) whole-rock composition, 2) mineral modal proportions, 3) mineral/melt partition coefficients, and 4) trapped melt fraction. We have used this method to test the hypothesis of closed system crystallization by estimating the composition of clinopyroxene, which is the last major phase in equilibrium with the interstitial melt. While whole-rock composition of Talavera and La Solanilla gabbros were determined by Orejana et al. (2009), in this work we have made a detailed estimation of the modal composition in the studied samples (see results in caption to Fig. 11). The mineral/melt partition coefficients used in this study have been extracted from experimental studies on basic magmas at thermodynamic conditions similar to those of the SCS gabbros (see references in caption to Fig. 12).

We have considered a 20% residual melt fraction, which represents a proportion coherent with the modal amount of minerals crystallised directly from the interstitial liquid. The modelled clinopyroxene composition is compared in Fig. 12 with the most primitive clinopyroxene from each sample (highest Cr contents). The pattern predicted by the equation of Bédard (1994) matches the composition of clinopyroxene from each sample, excepting LILE and LREE whose concentrations plot below or towards the lowest values of the represented clinopyroxene range (Fig. 12). These differences in LILE and LREE contents are difficult to reconcile with a closed magmatic system. The extremely high concentrations of elements such as Th and U in clinopyroxene (up to 2.86 ppm and 0.3 ppm in the Talavera gabbro; Fig. 7a, b) would imply dozens to hundreds ppm of Th in the liquid, even considering a wide range of mineral/basic melt partition coefficients (e.g. 0.06 to 0.01; Lemarchand et al., 1987; Hauri et al., 1994, respectively). Moreover, other minerals display remarkably high trace element concentrations (mainly REE), such as Na-rich plagioclase (Fig. 6g) and amphibole (Fig. 8e, f).



Fig. 11.- Bar diagram showing the relative proportion of selected trace elements in the main minerals from the Talavera and La Solanilla gabbros obtained by a mass balance calculation. Trace element composition of minerals used in the calculation considers averaged values (partially transformed clinopyroxene have not been considered). Modal composition used are those of Table 1: ol (7%), pl (54%), opx (16%), cpx (12%), amph (6%), phl (3%) and rest (2%) for Talavera sample and ol (15%), pl (58%), opx (6%), cpx (14%), amph (3%), phl (3%) and rest (1%) for La Solanilla sample

LILE and LREE enrichment in minerals (mainly clinopyroxene) from gabbroic rocks have been related with a process of contamination by an external evolved liquid (e.g. Tribuzio *et al.*, 1999a; Tiepolo and Tribuzio, 2005; Renna *et al.*, 2006). Figure 10b illustrate that the enrichment of the most incompatible REE in clinopyroxene (e.g., La) is associated to a higher degree of fractionation of LREE. High (La/Sm)_N values also correlate positively with Na₂O in plagioclase (Fig. 10a). This behaviour might be explained advocating the involvement of a LREE-rich external fluid/ melt. A recent study by Villaseca *et al.* (2011) focused on the geochronology and Hf isotope composition of zircons from the SCS gabbros, highlighted the heterogeneous *c*Hf values (+2.9 to -1.1) displayed by magmatic zircons from the Talavera gabbro and the presence of xenocrystic zircons with *c*Hf values typical of crustal rocks (-3 to -9) in the La Solanilla gabbro. These zircon data clearly support that both gabbroic bodies were contaminated to some degree either by partial melts derived from local assimilation of metamorphic wall-rocks or due to hybridization with coeval Variscan granite magmas as most studies suggest (e.g., Bea *et al.*, 1999; Villaseca *et al.*, 2011).

6.3. Nature of the parental magmas

Magmatism in collisional orogens is dominated by complex processes which commonly involve melting of contrasted sources (crust and mantle), fractionation, mixing of the generated magmas and local assimilation of country rocks. Furthermore, hybridization may occur at different crustal levels. Thus, the geochemical affinity of magmas parental to mantle-derived rocks is not always a straightforward issue.

The late-Variscan mafic rocks from central Spain develop well-characterised medium- to high-K calc-alkaline trends (Villaseca et al., 2004; Orejana et al., 2009; Scarrow et al., 2009), with the most primitive gabbros showing orthopyroxene (typically subalkaline) as a near liquidus phase. However, the presence of Ti-rich minerals in these rocks, such as kaersutite and Ti-phlogopite, have suggested an alkaline affinity (Molina et al., 2009), and the possibility that the calc-alkaline composition resulted from chemical interaction (hybridization) at depth between alkaline basic magmas and felsic melts, prior to the crystallization process (Scarrow et al., 2009). Kaersutite and pargasite are amphiboles typical of alkaline basic rocks and metasomatised mantle xenoliths and their formation occurs early during the crystallization process (e.g. Martin, 2007). Nevertheless, the presence of these minerals in subalkaline series is not an uncommon phenomenon. Féménias et al. (2006) described core to rim zoning patterns in amphiboles from basaltic to rhyolitic calc-alkaline dykes, including pargasite, kaersutite, edenite and magnesiohastingsite cores. A similar chemical evolution was found by Kawakatsu and Yamaguchi (1987) from brown magnesiohornblende and magnesiohastingsite towards green magnesiohorblende and actinolite in calc-alkaline plutonic rocks. In fact, pargasite displays a heterogeneous major element composition, with pargasite s.s. occurring almost exclusively in alkaline rocks and the tschermakitic pargasite being typical of calc-alkaline series (Ridolfi and Renzulli, 2012). The composition of pargasite from the Talavera and La Solanilla gabbros is similar to the tschermakitic pargasite described by Ridolfi and Renzulli (2012) and fits the chemical criteria proposed by these authors (mainly K in A-site below 0.22) apfu) for amphiboles formed from calc-alkaline magmas. Even kaersutite can precipitate as an interstitial mineral from tholeiitic magmas (e.g. Tribuzio et al., 2000; Coogan et al., 2001). These latter authors suggest that amphibole formation can be ascribed to the percolation of relatively evolved melts. Ti-rich biotite-phlogopite has also been described in subalkaline basic plutonic rocks (e.g., Tribuzio et al., 1999a; Costa



Fig. 12.- Primitive Mantle-normalised trace element abundances of clinopyroxenes calculated with the EDM method (Equilibrium Distribution Method; from Bédard, 1994) for (a) Talavera and (b) La Solanilla gabbros, assuming a trap melt fraction (TMF) of 20%. Whole rock composition extracted from Orejana et al. (2009). Modal composition as that of caption to Fig. 11, with the exception of clinopyroxene which has been balanced with the TMF in order to get a total of 100%. Partition coefficients used: olivine/melt from Bédard (1994); plagioclase/melt from Bindeman and Davis (2000) and Bédard (1994); ortopyroxene/melt from Green et al. (2000) and Kennedy et al. (1993) and clinopyroxene/melt from Hart and Dunn (1993), Hauri et al. (1994) and Adam and Green (2006). Trace element abundances of Cr-rich clinopyroxenes from each sample (grey field) are also reported for comparison. Normalising values after McDonough and Sun (1995).

et al., 2002; Claeson and Meurer, 2004). Thus, the presence of Ti-rich minerals can not be considered as an unambiguous evidence of crystallization from an alkaline melt.

Amphibole and phlogopite in the Talavera and La Solanilla gabbros are not abundant early precipitated phases, but latestage interstitial minerals present in low proportions (<9%), locally in disequilibrium with the surrounding phases (e.g. clinopyroxene; Fig. 2d, f). These textural and chemical features, taken together with the whole-rock chemistry, support a subalkaline affinity for the SCS Variscan basic melts. We propose that the extensive crystallization of initially H₂Osubsaturated magmas, in conjunction with the addition of external components during emplacement, might have led to subsequent enrichment in incompatible elements, including volatiles, and melt saturation in amphibole. The highly enriched trace element composition of most minerals in the studied gabbros (e.g. LREE in plagioclase, REE in orthopyroxene, clinopyroxene and amphibole) exclude the possibility of an initial tholeiitic magma, and are very similar to the composition of clinopyroxene and amphibole in other calcalkaline post-orogenic Variscan gabbros (e.g. the Corsica– Sardinia batholith; Renna *et al.*, 2006).

Apart from the late growth of Ti-rich minerals, no strong evidence against a dominant calc-alkaline affinity in the SCS gabbros is apparent. The last events of the Variscan orogeny are characterised in central Spain (and in many other western Variscan massifs) by a widespread felsic magmatism, with abundant granite intrusions emplaced in the range ~310-298 Ma (e.g., Orejana et al., 2012; Merino Martínez et al., 2014, and references therein). Late Variscan magmatism, including both the felsic magmas and the accompanying scarce basic melts, could have been originated by adiabatic decompression as lithosphere experienced considerable thinning during the orogen collapse (e.g., Orejana et al., 2009; Scarrow et al., 2009). Whole-rock geochemistry of the primitive olivinebearing gabbros, such as negative Nb-Ta and Ti anomalies, LILE enrichment and low Ta/Yb and Ce/Pb ratios, points to a calc-alkaline nature (Orejana et al., 2009). Furthermore, other basic rocks that intruded as large dyke swarms in the SCS in a later stage (~290 Ma; Galindo et al., 1994), show similar calc-alkaline fingerprints as those described above (Gb2 and Gb3 series in Villaseca et al., 2004). Accordingly, we think that a lithospheric mantle metasomatised by crustallike (collisional?) components would be the source of these post-collisional basic magmas.

The involvement of an alkaline component in the origin of the SCS primitive gabbros is not obvious. Their whole-rock composition is difficult to reconcile with a possible magma mixing process involving an alkaline basic melt and a felsic component. The concentration of some major and trace elements in the SCS gabbros do not plot in an intermediate position between that of a possible felsic melt (e.g., the SCS felsic Variscan magmatism) hybridized with an alkaline basic magma. Recent studies have highlighted the importance of crystal transfer as a mechanism capable of producing non-linear elemental correlations between felsic and mafic magmas (e.g., Ubide et al., 2014). However, the SCS gabbros display lower or overlapping contents in some minor and trace elements (Ti, Nb, Zr) with respect to granite composition. This feature implies a notable deviation from an alkaline composition and, accordingly, would mean a significant crystal transfer, if this process had been important during mixing. The primitive nature of the analysed olivine-bearing gabbros, together with the absence of felsic minerals showing disequilibrium textures, prevents significant mixing with coeval granite magmas and supports that the calc-alkaline signature in the SCS gabbros could be considered as a primary feature.

The only basic alkaline magmatic rocks outcropping in the SCS are lamprophyric and diabasic dykes (Orejana et al., 2008) of Permian age (ca. 264 Ma; e.g., Scarrow et al., 2006; Perini and Timmerman, 2008), which intrude about 35 Ma after the Variscan gabbros are emplaced (305-300 Ma; Villaseca et al., 2011), in a fairly different tectonic regime. Some of these alkaline dykes display isotopically depleted signatures (ϵ Nd ~ +7; Orejana *et al.*, 2008), which suggests that metasomatism in the source mantle likely occurred shortly before alkaline melt generation (Orejana et al., 2008). The Variscan basic magmatism at ca. 300 Ma would register a previous and different mantle source. This transformation of the lithospheric (and asthenospheric) mantle under central Spain has been associated with delamination of dense crustal blocks in the thickened Variscan collisional belt, which is supported by Sm-Nd isotopic signatures of pre-Variscan, Variscan and post-Variscan basic magmas (e.g., Villaseca et al., 2004; Orejana et al., 2008; Gutierrez-Alonso et al., 2011).

7. Conclusions

The Talavera and La Solanilla gabbroic intrusions display a heterogeneous mineral composition, including incompatible trace element-rich interstitial phases, such as amphibole, phlogopite and other accessory minerals (ilmenite, zircon, apatite, baddeleyite, titanite). Major element composition of the main minerals show a high degree of variation: for instance, plagioclase show zoning patterns ranging from Carich cores to Na-rich overgrowths and clinopyroxene zoning seems to be related to disequilibrium reactions, leading to replacement by amphibole at crystal rims.

Trace element contents in clinopyroxene from both gabbros towards a more evolved composition accounts for a progressive enrichment of the residual melt in incompatible elements. However, this tendency is clearly modified by coprecipitation of phases with high partition coefficients for specific elements, such as apatite (LREE), zircon (Zr, Hf, Th, U), baddeleyite (Zr, Hf, U), amphibole (Nb) and phlogopite (Ba, Nb, V). A simple mass balance calculation indicates that the distribution of most trace elements is controlled by minerals crystallised from the interstitial liquid (amphibole, phlogopite and accessory phases).

The use of the Equilibrium Distribution Method of Bédard (1994) highlights that the very high LILE and LREE contents in clinopyroxene from both gabbroic samples can not be explained exclusively by an in-situ crystallization process in a closed-system situation. Positive correlation between La/Sm normalised ratios and La and Na contents, in clinopyroxene and plagioclase respectively, serve as evidence of contamination with an external evolved liquid, likely associated with granite melt hybridization, as stated by previous isotopic studies.

Petrographic features (e.g., presence of orthopyroxene, late precipitation of amphibole) and mineral composition (e.g., trace element-enriched clinopyroxene and amphibole composition) in the primitive SCS gabbros, seem to favour an original calc-alkaline geochemical affinity, instead of alkaline or tholeiitic. The bulk rock composition of the SCS gabbros and possible felsic and basic alkaline magmas go against a possible mixing process in their origin. Available isotopic data from the literature also suggest that mantle metasomatism beneath central Spain, before the end of the Variscan orogeny, is more likely caused by continental crust components, probably associated with delamination of dense crustal blocks from the thickened Variscan collisional belt.

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Supplementary material

(available at http://revistas.ucm.es/index.php/JIGE/)

- Supplementary eTable 1.- Major element composition of minerals from Talavera and La Solanilla gabbros (Spanish Central System)
- Supplementary eTable 2.- Trace element composition of minerals from Talavera and La Solanilla gabbros (Spanish Central System)